**Vježba 5.**

**PROUČAVANJE IZOBARNE PROMJENE STANJA IDEALNOG PLINA**

**Zadaci:**

1. Istražiti ovisnost volumena i temperature uz stalni tlak i količinu plina.
2. Prikažite dobivene rezultate grafički u *V*, *t*  i *V*, T koordinatnom sustavu.
3. Matematički (formulom) formulirajte ovisnost volumena plina o temperaturi uz stalan tlak i količinu tvari.
4. Provedite diskusiju nakon obavljenog mjerenja.

 **Pribor:** staklena cjevčica s kapljicom žive, metar, termometar, čaša s vodom, kuhalo

* Skiciraj pokus!
* Hoće li će se pri promjeni temperature promijeniti volumen plina u cjevčici?

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* Hoće li će se pri promjeni temperature promijeniti tlak plina u cjevčici?

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* Možemo li pomoću ovog pribora istražiti ovisnost volumena o temperaturi uz stalan tlak i količinu plina? Opišite kako bi proveli mjerenja!

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* Napravite pet mjerenja i rezultate mjerenja upišite u tablicu:

|  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- |
| Redni broj mjerenja | Δ*V* /cm3 |  *V* /cm3 |  *t* / 0C |  *T* /K |  |
| 1. |  |  |  |  |  |
| 2. |  |  |  |  |  |
| 3. |  |  |  |  |  |
| 4. |  |  |  |  |  |
| 5. |  |  |  |  |  |

* Promatrajući rezultate iz tablice, što možete zaključiti o vezi volumena i temperature plina pri stalnom tlaku i količini plina?

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* Nacrtajte na milimetarskom papiru grafičku ovisnost volumena o temperaturi (*V*,*t*). Pri tome neka vam temperaturna os bude od -300 °C do 150 °C.
* Odredite i napišite pri kojoj bi temperaturi volumen bio 0 cm3.

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* Nacrtajte na milimetarskom papiru grafičku ovisnost volumena o termodinamičkoj temperaturi (*V*,*T*).
* Na temelju grafičkog prikaza volumena *V* o termodinamičkoj temperaturi *T* napišite matematičku ovisnost *V* o *T*.

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* Provjerite svoj odgovor na prethodno pitanje računanjem kvocijenta volumena *V* i temperature *T* za svako mjerenje i te kvocijente upišite u prazni stupac u tablici.

Napišite što te vrijednosti pokazuju.

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